

Mechanism of Photoformation of 1-Cyanobenzosemibullvalene

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Summary Deuterium labelling reveals that 1-cyanobenzosemibullvalene derived from the sensitized irradiation of 2-cyanobenzobarrelene is formed by a di- π -methane

rearrangement initiated by vinyl-vinylcyano-bridging and not by the benzo-vinylcyano-mechanism proposed by Houk.

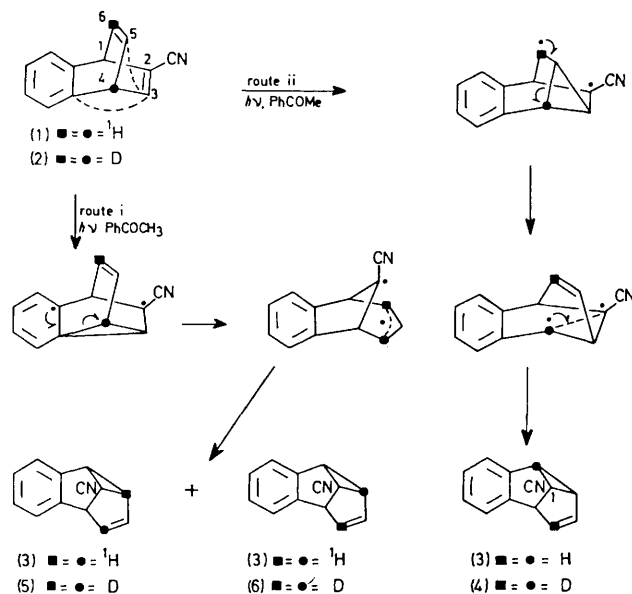
RECENTLY Houk reported a molecular orbital model to account for the direction of preferred di- π -methane bridging in benzonorbornadiene and benzobarrelene systems with polar substituents.¹ One case which was claimed to be in accord with the bonding predictions was that of 2-cyano-benzobarrelene (**1**) which gives 1-cyanobenzosemibullvalene (**3**) from the triplet-excited manifold.²

Although there are two *a priori* di- π -methane rearrangements capable of furnishing (**3**) (Scheme: route i initiated by benzo-vinylcyano-bridging, and route ii initiated by vinyl-vinylcyano-bridging) Houk invoked exclusively route i. In order to distinguish between these two alternatives we have examined the photochemistry of the deuterated mixture (**2**) of barrelenes.

The mixture (**2**), prepared by the addition of benzyne to *m*-deuteriobromobenzene, followed by treatment of the bromobarrelene with CuCN, consisted of an equal amount of two singly labelled isomers, one incorporating deuterium ($92 \pm 3\%$) at C-4 and the other deuterium ($92 \pm 3\%$) at C-6; within experimental error no deuterium isotope effects were discernable in the benzyne addition stage or in the subsequent photoisomerizations. Under conditions of triplet sensitization (450 W medium-pressure Hanovia lamp, Pyrex filter, acetophenone sensitizer, cyclohexane solution, nitrogen degassing) the barrelene mixture (**2**) gave only 1-cyanosemibullvalene (ϕ 0.36), the ¹H n.m.r. spectrum of which [τ (100 MHz, CCl₄) 2.66–3.05 (m, 4.00H, ArH), 4.34 (dd, 0.53 \pm 0.04H, 6-H, $J_{5,6}$ 2.5, $J_{6,7}$ 5.0 Hz), 4.78–4.87 (m, 0.97 \pm 0.04H, 7-H), 5.83 (br s, 1.02 \pm 0.03H, 5-H), 6.37 (d, 0.56 \pm 0.03H, 2-H, $J_{2,8}$ 7.0 Hz), and 6.67–6.78 (m, 1.03 \pm 0.04H, 8-H)] is in accordance solely with the presence of mixture (**4**), since the labels appear exclusively and equally at C-2 and C-6.

Possible scrambling of the label in the mixture (**4**), *e.g.* via a degenerate photochemical vinyl-cyclopropane rearrangement, was examined. Direct irradiation of (**4**) (Corex filter, 80% conversion) gave a deuterated mixture of 6-cyanobenzocyclo-octenes containing equal amounts of deuterium at C-5 and C-9 (ϕ 0.13), but the distribution of the label in recovered (**4**) was unchanged. Irradiation with acetophenone or acetone as sensitizer results in the disappearance of (**4**) ($\phi_{\text{dis}} < 0.05$) to give unidentified products,

but there was no change in the labelling pattern in (**4**) after 85% consumption. Thus the example reported in ref. 3 remains the only case of a semibullvalene observed to undergo a photochemical vinyl-cyclopropane rearrangement.³



SCHEME. Individual structures represent a 50:50 two-component mixture, each singly labelled with deuterium at the positions designated by ■ and ●.

Therefore Houk's account of the initial bonding process (route i) involved in the di- π -methane rearrangement of (**1**) is not in accord with our findings.

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¹ C. Santiago and K. N. Houk, *J. Amer. Chem. Soc.*, 1976, **98**, 3380.

² C. O. Bender and S. S. Shugarman, *J.C.S. Chem. Comm.*, 1974, 934.

³ C. O. Bender and J. Wilson, *Helv. Chim. Acta*, 1976, **59**, 1469.